

Report

Click Chemistry Approach to Assembly Proline Mimetic Libraries Containing 1,4-Substituted 1,2,3-Triazoles

Jose# E. Rodri#guez-Borges, Sofia Gonc#alves, Maria Lui#sa do Vale, Xerardo Garci#a-Mera, Alberto Coelho, and Eddy Sotelo

J. Comb. Chem., 2008, 10 (3), 372-375• DOI: 10.1021/cc800035z • Publication Date (Web): 30 April 2008

Downloaded from http://pubs.acs.org on March 25, 2009



R¹ = Me, ^tBu, Cy, R² = Bn, CH2-COOEt, CH2-S-Ph

More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 1 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML



Click Chemistry Approach to Assembly Proline Mimetic Libraries Containing 1,4-Substituted 1,2,3-Triazoles

José E. Rodríguez-Borges,^{*,†} Sofia Gonçalves,[†] Maria Luísa do Vale,[†] Xerardo García-Mera,[‡] Alberto Coelho,[§] and Eddy Sotelo^{‡,§}

CIQ-Departamento de Química, Faculdade de Ciências, Universidade do Porto, Rua do Campo Alegre, 687, 4169-007 Porto, Portugal, and Departamento de Química Orgánica and Instituto de Farmacia Industrial, Facultade de Farmacia, Universidade de Santiago de Compostela, Santiago de Compostela, 15782 Spain

Received March 3, 2008

The development of new therapeutic agents, as well as the identification of molecular probes for the study of the chemical/biological interfaces, is one of the major goals in biomedical research. In this context, the availability of large libraries of small organic molecules, covering as much chemical space as possible, is seen as the only means which guarantees potential modulation of the many biological targets that are ultimately being unveiled by genomics.¹ Therefore, advances in drug discovery depend heavily on the availability of synthetic transformations that allow the rapid assembly of complex molecular frameworks providing maximum diversity.

In this challenging scenario, cycloadditions² occupy an outstanding position because of their versatility, exploratory power, high degree of atom economy, and ability to generate small highly polyfunctionalized molecular skeletons. Within the diverse transformations comprising cycloadditions, aza-Diels-Alder reactions have attracted much interest, especially in those employing cyclopentadiene as starting material.³ On the other hand, not only the enormous potential of 2-azabicyclo[2.2.1]hept-5-enes as synthetic precursors of diverse valuable structures⁴ but also the growing biomedical interest in this template⁵ because of its easse of origination of proline mimetic structures has stimulated continuous advances in the development of synthetic methods⁶ able to produce functionally and stereochemically diverse 2-azabicyclo[2.2.1]hept-5-ene derivatives. Nevertheless, there is an increasing need for simple and efficient procedures that allow to incorporate unexplored diversity patterns in the bicyclic scaffold and simultaneously facilitate the assembly of more complex structures (e.g., by linking the 2-azabicyclo[2.2.1]hept-5-ene system to other well documented privileged druglike molecules or natural products).

In the context of our ongoing efforts toward the development of new methodologies to access structurally diverse heterocyclic libraries,⁷ we report herein the facile assembly of a library of 2-azabicyclo[2,2,1]hept-5-enes incorporating, at position 2 of the bicyclic skeleton, a 1,2,3-triazole nucleus linked through alkyl chains of variable lengths distances (Figure 1). The proposed synthetic pathway is based on the combination of two highly versatile transformations: the aza-Diels—Alder reaction and the Cu^I-catalyzed Huisgen cycloaddition (Figure 1). The libraries are not only novel but also exemplify a simple and highly efficient assembly of pharmacologically valuable triazole containing proline mimetic structures.



Figure 1. General structure of targeted libraries and retrosynthetic scheme.

The general structure of the targeted library, as well as the retrosynthetic analysis are presented in Figure 1. It should be pointed out that although Figure 1 discloses structures incorporating different alkylic linkages (n = 1, 2, or 3) between the privileged scaffolds, in this preliminary communication, we only describe the results obtained during the proof of the concept of the pathway, for example, using propargylamine as nitrogenated component (n = 1).

The feasibility of the proposed sequence relied on the successful preparation of the 2-azabicyclo[2.2.1]hept-5-ene template incorporating a reactive terminal acetylenic moiety at position 2. Thus, we proceeded to study the aza-Diels-Alder reaction of cyclopentadiene with a subset of glyoxylates 1a-c (or their synthetic equivalents, e.g., hemiacetals or hydrates) having different alkyl residues in the ester moiety [e.g., Me, t-Bu, Cy (Cy = cyclohexyl)] and employing propargylamine as the amine component (Scheme 1). The three-component reaction that takes place in a consecutive manner is highly accelerated by the addition of a Lewis acid (e.g., boron trifluoride). The transformation starts with the in situ generation of a highly reactive azadienophile (iminium cation)³ which rapidly undergoes a [4 + 2] cycloaddition under mild conditions to afford the desired (±) 2-propargyl-2-azabicyclo[2.2.1]hept-5-ene-3carboxylates (2a-e) as variable endo/exo mixtures that were chromatographically separated under standard conditions.^{8,9} Table 1 shows the isolated yield for all the six 2-azabicycloalkene adducts prepared 2a-f.¹⁰

The stereochemistry of the exo and endo adducts was unequivocally determined through NMR spectral data (¹H NMR, ¹³C NMR, DEPT, HMQC, HMBC)⁹ and, additionally, for adducts **2a** and **2b** through 1 D NOE NMR experiments⁹ at room temperature. The NOE effects observed between 3-H and the corresponding protons in the close vicinity, shown in Figure 2, allow an unambiguous attribution of the referred structures. The observed exo/endo ratio (~3:1) could be

^{*} To whom correspondence should be addressed. E-mail: jrborges@fc.up.pt.

[†] CIQ-Departamento de Química, Universidade do Porto.

^{*} Departamento de Química Orgánica, Universidade de Santiago de Compostela.

[§] Instituto de Farmacia Industrial, Universidade de Santiago de Compostela.



Figure 2. Assignment of the stereochemistry of exo/endo adducts based on 1 D NOE NMR.





Table 1. Isolated Yields of (\pm) -2-Propargyl-2-azabicyclo[2.2.1]hept-5-ene-3-carboxylates 2

		yield (%)		
compound	\mathbb{R}^1	exo	endo	
2a	Me	68		
2b			20	
2c	'Bu	53		
2d	G	~~	18	
2e	Cy	22	15	
21			15	

explained considering that in the close vicinity of the C=N bond of the iminium ion (E configuration) the methylene group (Csp³) exerts a larger steric hindrance than the ester moiety (Csp²). Consequently, to minimize stereochemical interactions between the methylene group of the diene and the propargyl group, the approach diene-dienophile must occur in an exo manner. That is to say that the stereochemical factors are more important than secondary orbital interactions between the π -systems of cyclopentadiene and the ester group in the dienophile.

Once a set of exo and endo (\pm) -2-propargyl-2azabicyclo[2.2.1]hept-5-ene-3-carboxylates 2a-g was available, the structural decoration/modification of the bicyclic scaffold, through construction of the 1,2,3-triazole nucleus, was initiated. For this purpose we made use of the well established copper-catalyzed Huisgen 1,3-dipolar cycloaddition of azides and terminal alkynes,¹⁰ to date the most reliable click reaction.¹¹ Its applications in numerous areas has highlighted the value of its mild reaction conditions, enabling the practical and efficient preparation of 1,4disubstituted 1,2,3-triazoles from an unprecedent range of substrates, with excellent selectivity.^{10d} The Cu^I-catalyzed Huisgen reaction has emerged as a tailored transformation to either undertake the structural diversification of privileged scaffolds¹² or establish linkages between different molecules.13

The first stage of the study involved a comprehensive screening process, in order to identify a mild and efficient procedure able to assemble the 1,2,3-triazole ring, within the plethora of published protocols to perform the Cu^I-catalyzed Huisgen reaction.^{10d,12,13} In a pilot experiment, the reaction

Scheme 2. Synthesis of 1,2,3-Triazole-2azabicyclo[2.2.1]hept-5-ene-3-carboxylate Conjugates under the Cu^I-Catalyzed Huisgen Cycloaddition



R¹ = Me, ^tBu, Cy, R² = Bn, CH2-COOEt, CH2-S-Ph

 Table 2.
 Isolated Yields of 1,2,3-Triazole-2

 azabicyclo[2.2.1]hept-5-ene-3-carboxylate Conjugates

compound		R^1	yield (%)	
	\mathbb{R}^2		exo	endo
4 a	Bn	Me	86	
4b	Bn	Me		79
4c	Bn	'Bu	81	
4d	Bn	'Bu		72
4 e	Bn	Cy	83	
4f	Bn	Ċv		73
4g	CH ₂ -COOEt	Me	74	
4h	CH ₂ -COOEt	Me		63
4i	CH ₂ -COOEt	'Bu	68	
4i	CH ₂ -COOEt	'Bu		61
4k	CH ₂ -COOEt	Cy	76	
41	CH ₂ -COOEt	Ċy		68
4m	CH ₂ -S-Ph	Me	78	
4n	CH ₂ -S-Ph	Me		67
4o	CH ₂ -S-Ph	'Bu	64	
4p	CH ₂ -S-Ph	'Bu		54
4g	CH ₂ -S-Ph	Cy	68	
4r	CH ₂ -S-Ph	Ċv		56

of benzyl azide (3a) with the exo (2a) and endo (2b) adducts of the methyl ester derivative was employed as a model system. Systematic evaluation of the optimum experimental conditions to achieve the proposed transformation included the screening of diverse catalytic copper species (CuCl, CuBr, Cu(0), Cu(OAc)₂, CuSO₄), solvents (THF, *i*-PrOH-H₂O, DMF, t-Bu-H₂O) and other additives (e.g., TEA, DIPEA, or sodium ascorbate). These preliminary experiments evidenced the superiority of the CuSO₄/sodium ascorbate system to perform the Huisgen [3 + 2]-cycloaddition on the polyfunctional frameworks 2. These optimized conditions were successfully applied to the reaction of adducts 2 with the azides (3a-c) selected for method development. Reactions occurred at room temperature and were usually finished in less than 3 h, generating a library of 18 new 1,2,3-triazole-2-azabicyclo[2.2.1]hept-5-ene-3-carboxylate conjugates 4 (Scheme 2, Table 2).^{9,14}

As observed (Table 2), independently of the structural differences within the azide precursors (3) and the alkoxy residue in the ester moiety of the 2-azabicycle (2a-e), the yield of the 1,2,3-triazole derivatives obtained from the exo

adducts is slightly superior ($\sim 10\%$) to those prepared starting form the endo adducts. A possible explanation for this finding may be related to the instability of the endo adducts, which suffer a slow degradation under experimental conditions.

In summary, a simple and efficient procedure allowing the rapid assembly of libraries of pharmacologically valuable 1,2,3-triazoles, containing proline mimetic structures, has been developed. Furthermore, the potential of the combination of two highly versatile cycloadditions (e.g., aza-Diels–Alder and the Cu^I-catalyzed Huisgen reaction) as a successful strategy to explore new diversity space and address the structural decoration of privileged scaffolds has been established. Further studies to extend these procedures to homopropargylamines as well as other cyclic dienes and glyoxylates, are currently in progress in our laboratories.

Acknowledgment. This work was financially supported by the Galician Government (project: 07CSA008203PR) and by the Portuguese Government (FCT-Pluri-annual and Programmatic Funding-POCTI/QUI/44471/2002). E.S. and A.C. are researchers of the Isidro Parga Pondal program (Xunta de Galicia, Spain). M.C.V. and J.E.R.-B. are researchers at CIQ-Portugal.

Supporting Information Available. Detailed experimental procedures, spectroscopic data, and copies of NMR and mass spectra for all compounds described. This information is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

- (a) Bartlett, P. A.; Entzeroth, M. Exploiting Chemical Diversity for Drug Discovery; RSC Publishing: Cambridge, U.K., 2006.
 (b) Verheij, H.; Robeson, B. L. Genomic/Proteomic Technol.
 2002, 2, 34–35. (c) Thomas, G. L.; Wyatt, E. E.; Spring, D. R. Curr. Opin. Drug Discovery Dev. 2006, 9, 700–712. (d) Gordon, E. M.; Kerwin, J. F. Combinatorial Chemistry and Molecular Diversity in Drug Discovery; Wiley-VCH: Weinheim, Germany, 1998. (e) Nicolau, K. C.; Hanko, R.; Hartwing, W. Handbook of Combinatorial Chemistry; Wiley-VCH: Weinheim, Germany, 2002.
- (2) (a) Dennis, N. Org. React. Mech. 2007, 427–467. (b) Battiste, M. A.; Pelphrey, P. M.; Wright, D. L. Chem.—Eur. J. 2006, 12 (13), 3438–3447. (c) Ess, D. H.; Jones, G. O.; Houk, K. N. Adv. Synth. Catal. 2006, 348, 2337–2361. (d) Domingo, L. R. Mini-Rev. Org. Chem. 2005, 2 (1), 47–57.
- (3) (a) Boger, D. L.; Weinreb, S. M. In *Hetero Diels-Alder Methodology in Organic Synthesis*; Organic Chemistry Monograph 47; Wasserman, H. H., Ed.; Academic Press: New York, 1987. (b) Weinreb, S. M. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I.; Paquette, L. A., Eds.; Pergamon Press: Oxford, U.K, 1991; Vol. 5, pp 401–449. (c) Rowland, G. B.; Rowland, E. B.; Zhang, Q.; Antilla, J. C. *Curr. Org. Chem.* 2006, *10* (9), 981–1005. (d) Vale, M. L. C.; Rodríguez-Borges, J. E.; Caamaño, O.; Fernández, F.; García-Mera, X. *Tetrahedron* 2006, *62*, 9475–9482, and references therein. (e) Rodriguez-Borges, J. E.; G.-Mera, X.; Fernandez, F.; Lopes, V. H. C.; Magalhães, A. L.; Cordeiro, M. N. D. S. *Tetrahedron* 2005, *61*, 10951–10957, and references therein. (f) Fernandez, F.; G.-Mera, X.; Vale, M. L. C.; Rodriguez-Borges, J. E. *Synlett* 2005, *2*, 319–321, and references therein.
- (4) For representative examples see: (a) Maison, W. Eur. J. Org. Chem. 2007, 2276–2284. (b) Roberts, S. M.; Smith, C.; Thomas, R. J. J. Chem. Soc., Perkin Trans. 1 1990, 1493– 1495. (c) Browser, A. M.; Madalengoitia, J. S. Tetrahedron

Lett. 2005, 46, 2896–2872. (d) Fray, A. H.; Augeri, D. J.; Kleinman, E. F. J. Org. Chem. 1988, 53, 896–899. (e) Maison, W.; Kuntzer, D.; Grohs, D. Synlett 2002, 1795–1798. (f) Liu, Z.; Rainier, J. D. Org. Lett. 2006, 8, 459–462. (g) Ward, S. E.; Holmes, A. B.; McCague, R. Chem. Commun. 1997, 2085– 2086.

- (5) (a) De Diego, S. A.; Muñoz, P.; González-Muñiz, R.; Herranz, R.; Martín-Martínez, M.; Cenarruzabeitia, E.; Frechilla, D.; Del Rio, J.; Jimeno, M. L.; García-López, T. *Bioorg. Med. Chem.* 2005, *15*, 2279–2283. (b) Harris, P. W. R.; Brimble, M. A.; Muir, V. J.; Lai, M. Y. H.; Trotter, N. S.; Callis, D. J. *Tetrahedron* 2005, *61*, 10018–10035.
- (6) (a) Tararov, V. I.; Kadyrov, R.; Kadyrova, Z.; Dubrovina, N.; Börner, A. *Tetrahedron Asymm.* 2002, *13*, 25–28. (b) Södergren, M. J.; Andersson, P. G. *Tetrahedron Lett.* 1996, *37*, 7577–7580. (c) Mellor, J. M.; Richards, N. G. J.; Sargood, K. J.; Anderson, D. W.; Chamberlin, S. G.; Davies, D. E. *Tetrahedron Lett.* 1995, *36*, 6765–6768.
- (7) (a) Coelho, A.; Sotelo, E. J. Comb. Chem. 2005, 7, 526–529.
 (b) Coelho, A.; Sotelo, E. J. Comb. Chem. 2006, 8, 388–400.
- (8) General procedure for the synthesis of N-propargyl-2azabicyclo[2,2,1]hept-5-ene-2-carboxylates 2a-f: A solution of propargyl amine (3.20 mL, 47 mmol) in dry CH₂Cl₂ (30 mL) was added under argon to a stirred suspension of glyoxylate/hydrate (47 mmol) and 3 Å molecular sieves (29 g) in dry CH₂Cl₂ (140 mL) at 0°C. When the addition was complete the reaction mixture was cooled to-78°C and treated successively with trifluoroacetic acid (3.60 mL, 47 mmol), boron trifluoride etherate (5.90 mL, 47 mmol), and freshly distilled cyclopentadiene (7.5 mL, 91 mmol). After 6 h, saturated aqueous NaHCO₃ solution (120 mL) and then solid NaHCO₃ (12 g) were added. The reaction mixture was allowed to reach room temperature and filtered through a pad of Celite, and the organic layer of the resulting mixture was separated. The aqueous layer was extracted with CH_2Cl_2 (3 × 100 mL). The pooled organic layers were washed with saturated aq NaHCO₃ solution ($3 \times 100 \text{ mL}$) and brine ($3 \times 100 \text{ mL}$) and were dried over Na₂SO₄. Removal of the solvent on a rotary evaporator left an oily residue that upon chromatography on silica gel using a (19:1) hexane/EtOAc mixture as eluent afforded pure exo-adduct in the early fractions and the endoadduct in the latter ones, both as colourless oils.
- (9) Complete description of the spectroscopic and analytical data of all compounds described is included in the Supporting Information.
- (10) (a) Huisgen, R. Angew. Chem., Int. Ed. 1963, 2, 565–598.
 (b) Huisgen, R. Angew. Chem., Int. Ed. 1963, 2, 633–645.
 (c) Rostovtset, V. V.; Green, L. G.; Fokin, K. B.; Sharpless, B. Angew. Chem., Int. Ed. 2002, 41, 2596–2599. (d) Tornoe, C. W.; Christensen, C.; Medal, M. J. Org. Chem. 2002, 67, 3057–3064. (e) Bock, V. D.; Hiemstra, H.; van Maarseveen, J. H. Eur. J. Org. Chem. 2006, 51–68.
- (11) (a) Kolb, H. C.; Finn, M. G.; Sharpless, B. Angew. Chem., Int. Ed. 2001, 40, 2004–2021. (b) Kolb, H. C.; Sharpless, B. Drug Discovery Today 2003, 8, 1128–1137.
- (12) For representative examples, see: (a) Sivakumar, K.; Xie, K.; Cash, B. M.; Long, S.; Barnhill, H. N. Org. Lett. 2004, 6, 4603–4606. (b) Suárez, P. L.; Gándara, Z.; Gómez, G.; Fall, Y. Tetrahedron Lett. 2004, 45, 4619–4621. (c) Kuijpers, B. H. M.; Groothuys, S.; Keereweer, A. R.; Quaedflieg, P. J. L. M.; Blaauw, R. H.; Van Delfit, F. L.; Rutjes, F. P. J. T. Org. Lett. 2004, 6, 3123–3126. (d) Reck, F.; Zhou, F.; Girardot, M.; Kern, G.; Eyermann, C. J.; Hales, N. J.; Ramsey, R. R.; Gravestock, M. B. J. Med. Chem. 2005, 48, 499–506. (e) Khanetskyy, B.; Dallinger, D.; Kappe, C. O. J. Comb. Chem. 2004, 6, 884–892.
- (13) For representative examples, see: (a) Whiting, M.; Muldoon, J.; Lin, Y-C.; Silverman, S. M.; Lindstrom, W.; Olson, A. J.; Kolb, H. C.; Finn, M. G.; Sharpless, B.; Elder, J. H.; Fokin, V. V. Angew. Chem., Int. Ed. 2006, 45, 1435–1439. (b) Wu, P.; Feldman, A. K.; Nuget, A. K.; Hawker, C. J.; Scheel, A.;

Voit, B.; Pyun, J.; Fréchet, J. M. J.; Sharpless, B.; Fokin, V. V. Angew. Chem., Int. Ed. 2004, 43, 3928–3932. (c) Wolfbeis,
O. S. Angew. Chem., Int. Ed. 2007, 46, 2980–2982. (d) Lewis,
W. G.; Green, L. G.; Grynszpan, F.; Radic, Z.; Carlier, P. R.; Taylor, P.; Finn, M. G.; Sharpless, B. Angew. Chem., Int. Ed. 2002, 41, 1053–1057. (e) Zhang, J.; Chen, H-N.; Chiang, F. I.; Takemoto, J. Y.; Bensaci, M.; Chang, C.-W. T. J. Comb. Chem. 2007, 9, 17–19. (f) Aucagne, V.; Leigh, D. A. Org. Lett. 2006, 8, 4505–4507. (g) Pisaneschi, F.; Cordero, F. M.; Lumini, M.; Brandi, A. Synlett 2007, 2882–2884.

(14) General procedure for the Cu-catalyzed Huisgen cycloaddition between *N*-propargyl-2-azabicyclo[2,2,1]-5-ene-2-carboxylates (2a-f) and azides (3a-c): To a Kimble vial containing a solution of the *N*-propargyl-2-azabicyclo[2,2,1]hept-5-ene-2-

Journal of Combinatorial Chemistry, 2008 Vol. 10, No. 3 375

carboxylates $2\mathbf{a}-\mathbf{e}$ (0.38 mmol) in 3 mL of a THF/H₂O (5: 1), a mixture of the corresponding azide ($3\mathbf{a}-\mathbf{c}$) (0.38 mmol), sodium ascorbate (0.08 mmol), and CuSO₄ • 5H₂O (0.04 mmol) were added. The reaction mixture was orbitally stirred, at 40°C, until reactions reached completion (2–5 h), filtered through a Celite pad, and successively washed (5 mL) with THF, CH₂Cl₂, MeOH, and AcOEt. Evaporation of the solvents afforded an oily residue that was diluted with CH₂Cl₂ (10 mL), treated with PS-TMG (500 mg), and stirred for 0.5 h to remove the copper salts. Isolation of the desired 1,2,3-triazoles $4\mathbf{a}-\mathbf{r}$ was accomplished by preparative chromatography.

CC800035Z